

## 14.1

## HYDROGEN IONS AND ACIDITY

## Section Overview

## Objectives

1. Identify conjugate acid-base pairs of ions, acids, or hydroxides and an appropriate equilibrium.
2. Calculate hydroxide and hydronium ion concentrations from  $K_b$  and  $K_a$  values and the concentration of a weak acid or base.
3. Calculate the pH of weak acid solutions.

## Vocabulary

- |                   |                             |                   |
|-------------------|-----------------------------|-------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. weak acid      |
| 4. autoionization | 5. hydronium ion            | 6. autoionization |
|                   | 7. weak base                | 8. pH             |

## Key Equations

- |  |   |
|--|---|
| 1. $K_a = \frac{[\text{H}_3\text{O}^+][\text{A}^-]}{[\text{HA}]}$ (Eq. 14.1) | 2. $K_b = \frac{[\text{B}][\text{H}_3\text{O}^+]}{[\text{BH}^+]}$ (Eq. 14.2)    |
| 3. $\text{pH} = -\log[\text{H}_3\text{O}^+]$ (Eq. 14.3)                      | 4. $\text{pH} = \text{p}K_a + \log \frac{[\text{A}^-]}{[\text{HA}]}$ (Eq. 14.4) |

## Part A: Completion

Use the appropriate number of significant figures in the answers to the following questions. Show calculations for all problems. Answers should be written in the space provided.

1. What products are **1**  **2**  formed by the autoionization of water? Write the autoionization reaction of water which produces an hydronium ion and a hydroxide ion.
2. Hydrochloric acid dissociates from **1**  **2**  in water. Write the **3**  **4**  autoionization reaction of the hydrochloric acid. Write the **5**  **6**  **7**  **8**  **9**  **10**  **11**  **12**  **13**  **14**  **15**  **16**  **17**  **18**  **19**  **20**  **21**  **22**  **23**  **24**  **25**  **26**  **27**  **28**  **29**  **30**  **31**  **32**  **33**  **34**  **35**  **36**  **37**  **38**  **39**  **40**  **41**  **42**  **43**  **44**  **45**  **46**  **47**  **48**  **49**  **50**  **51**  **52**  **53**  **54**  **55**  **56**  **57**  **58**  **59**  **60**  **61**  **62**  **63**  **64**  **65**  **66**  **67**  **68**  **69**  **70**  **71**  **72**  **73**  **74**  **75**  **76**  **77**  **78**  **79**  **80**  **81**  **82**  **83**  **84**  **85**  **86**  **87**  **88**  **89**  **90**  **91**  **92**  **93**  **94**  **95**  **96**  **97**  **98**  **99**  **100**  **101**  **102**  **103**  **104**  **105**  **106**  **107**  **108**  **109**  **110**  **111**  **112**  **113**  **114**  **115**  **116**  **117**  **118**  **119**  **120**  **121**  **122**  **123**  **124**  **125**  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# 19 2 Hydrogen Ions And Acidity Section Review Answers

**Chao Zhang**



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