

**atomic power**

A popular synonym for *nuclear power*. (See *atomic energy, nuclear energy*)

**atomic ratio**

Ratio of quantities of different substances in terms of number of atoms of each.

**atomic stopping power**

(See *stopping power*)

**atomic weight, (awu)**

The weighted mean of the masses of the neutral atoms of an element expressed in atomic weight units. Unless otherwise specified it refers to a naturally occurring form of the element. The **atomic weight unit, awu**, is exactly one-sixteenth of the weighted mean of the masses of the neutral atoms of oxygen of isotopic composition found in fresh lake or rain water;  $1 \text{ awu} = 1.660 \times 10^{-24} \text{ gm} = 1.000272 \text{ amu}$ . The experimentally determined ratio of the atomic weight unit to the *atomic mass unit* is called the **atomic mass conversion factor**. The value of this factor is uncertain in the seventh significant figure because of variations in the isotopic composition of oxygen from different natural sources. In the future its value, and hence that of the atomic weight unit, may be rendered exact by definition. (See *atomic mass*)

**a.w., chemical scale** is the average weight of the atoms of an element (of isotope distribution found in nature) referred to the weight of exactly 16 for the average weight of the atoms of oxygen of isotope distribution found in fresh lake or rain water. The value may be computed from atomic weights on the physical scale by division by the factor 1.000272.

**a.w., physical scale** is the weight of an atom or the average weight of a mixture of atoms of an element referred to the exact value 16 for the

principle isotope of oxygen,  $O^{16}$ . Atomic weights on the physical scale are expressed in atomic mass units, amu. They may be computed from the atomic weights on the chemical scale by multiplication by the factor 1.000272.

**attenuation**

1. (Radiation theory) The reduction in the flux density, or power per unit area, with distance from the source; it may be due to absorption, to scattering, or to both processes.
2. (Nuclear physics) The reduction in the intensity of radiation upon passage through matter; in general, it is due to a combination of scattering and absorption. (*particle attenuation, energy attenuation*)

The term is best used in a sense that excludes the geometric decrease of intensity with distance from the source (inverse-square effect); the attenuation then depends only on the nature of the radiation and of the material traversed.

**attenuation factor**

A measure of the *opacity* of a layer of material for radiation traversing it; the ratio of the incident intensity to the transmitted intensity. It is equal to  $I_0/I$ , where  $I_0$  and  $I$  are the intensities of the incident and emergent radiation, respectively. In the usual sense of *exponential absorption* ( $I = I_0 e^{-\mu x}$ ), the attenuation factor is  $e^{+\mu x}$ , where  $x$  is the thickness of the material and  $\mu$  is the *absorption coefficient*.

**Auger effect**

A nonradiative transition of an atom from an excited electronic energy state to a lower state with the emission of an electron. The term usually refers to the x-ray region of energy states. The final state corresponds to one higher degree of ionization

# Glossary Of Terms In Nuclear Science And Technology

**J Rink**



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