

Half Reaction

Standard Potential (V)

F_2	+	$2e^-$	\rightleftharpoons	$2F^-$	+2.87
Pb^{4+}	+	$2e^-$	\rightleftharpoons	Pb^{2+}	+1.67
Cl_2	+	$2e^-$	\rightleftharpoons	$2Cl^-$	+1.36
O_2	$+ 4H^+$	$4e^-$	\rightleftharpoons	$2H_2O$	+1.23
Ag^+	+	$1e^-$	\rightleftharpoons	Ag	+0.80
Fe^{3+}	+	$1e^-$	\rightleftharpoons	Fe^{2+}	+0.77
Cu^{2+}	+	$2e^-$	\rightleftharpoons	Cu	+0.34
$2H^+$	+	$2e^-$	\rightleftharpoons	H_2	0.00
Pb^{2+}	+	$2e^-$	\rightleftharpoons	Pb	-0.13
Fe^{2+}	+	$2e^-$	\rightleftharpoons	Fe	-0.44
Zn^{2+}	+	$2e^-$	\rightleftharpoons	Zn	-0.76
Al^{3+}	+	$3e^-$	\rightleftharpoons	Al	-1.66
Mg^{2+}	+	$2e^-$	\rightleftharpoons	Mg	-2.36
Li^+	+	$1e^-$	\rightleftharpoons	Li	-3.05

Electro Chemical Science

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