

# Chemistry Study Guide

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## 🎯 Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 📖

## 💡 Matter:

Anything that has **mass** and **occupies space** 🌌🌍

Examples: air, gases, liquids, solidz!

## 🔬 Atoms:

Atoms are the **smallest, indivisible particle** 📏👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## 📏 Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	$p^+$	1	+1	Nucleus
Neutron	$n$	1	0	Nucleus
Electron	$e^-$	$\frac{1}{1836}$	-1	Orbit/Shell

# Section 2 nuclear Chemistry Study Guide Key

**Zhe Xu**



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