

Rate of Change (Slope) STAR Practice

$$\text{Slope (Rate of Change) of a line} = \frac{\text{rise}}{\text{run}} = \frac{y_2 - y_1}{x_2 - x_1}$$

1. What is the slope of the line that passes through the points $(-6, -2)$ and $(2, 7)$?

- A. $-\frac{9}{8}$
 B. $\frac{9}{8}$
 C. $2\frac{3}{4}$
 D. $-\frac{9}{8}$

2. What is the slope of the line that passes through the points $(3, -2)$ and $(7, 0)$?

- A. 2
 B. $\frac{1}{2}$
 C. $2\frac{1}{2}$
 D. $\frac{1}{2}$

3. The slope of a line that passes through the points $(3, a)$ and $(7, 20)$ is 4. What is the value of a ?

- A. 10
 B. 1
 C. 4
 D. 8

4. What is the rate of change of the following graph?



- A. 2
 B. $\frac{1}{2}$
 C. $-\frac{1}{2}$
 D. $-\frac{3}{2}$

5. The graph models the linear relationship between the position of a ball in the air and the seconds that have elapsed since the ball was thrown. Which of the points best represents the rate of change in position with respect to time?

- A. (0, 0) B. (2, 2)
 C. (2, 0) D. (0, 2)



Multiple Choice Rate Of Change Questions

Wolfgang Guggemos



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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the

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