

- d. Nope, you will have to ask our petty cash custodian for your reimbursement. All disbursements below ₱5,000 will have to be paid out of the petty cash fund.
8. By requiring that all deposits must be deposited intact and all disbursements, except petty disbursements, be made through check, you are implementing an internal control called
- Regular bank reconciliations.
  - Imprest system.
  - Voucher system.
  - Bossy system.
9. By requiring that all disbursements must be supported by CDVs, you are implementing an internal control called
- Regular bank reconciliations.
  - Imprest system.
  - Voucher system.
  - Bossy system.
10. Which of the following is incorrect regarding the petty cash fund?
- The petty cash fund should only be used for recurring expenses that are not material enough to warrant disbursement through check.
  - Each petty cash disbursement must be supported by a properly authorized petty cash voucher.
  - The requesting party should affix his or her signature on the petty cash voucher acknowledging the receipt of the petty cash.
  - For a big company, each petty cash voucher must be authorized by the Chief Executive Officer or President.

## Chapter 9

### Accounting Practice Set

#### Learning Competencies

The learners should be able to.....

- Perform the steps in the accounting cycle, from preparation of documents to the preparation, analysis, and interpretation of financial statements

#### Practice Set

The November 30, 20x1 trial balance of Rak N' Koll Co., a call center business, shows the following information:

ACCOUNTS	Debit	Credit
Cash	340,000	
Accounts receivable	-	
Prepaid insurance	-	
Computer equipment	1,400,000	
Accumulated depreciation		240,000
Loans payable		1,000,000
Owner's equity		1,254,000
Owner's drawings	3,300,000	
Service fees		4,890,000
Salaries expense	990,000	
Rent expense	1,100,000	
Utilities expense	165,000	
Office supplies expense	84,000	
Insurance expense	5,000	
<b>Totals</b>	<b>7,384,000</b>	<b>7,384,000</b>

# Accounting Practice Set Solutions

**N Colangelo**



**Accounting Practice Set Solutions:**

## Enjoying the Track of Appearance: An Emotional Symphony within **Accounting Practice Set Solutions**

In some sort of consumed by monitors and the ceaseless chatter of immediate communication, the melodic elegance and psychological symphony created by the published term frequently diminish into the backdrop, eclipsed by the persistent noise and disturbances that permeate our lives. However, located within the pages of **Accounting Practice Set Solutions** a wonderful fictional prize full of fresh emotions, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, that fascinating masterpiece conducts visitors on a psychological trip, skillfully unraveling the hidden tunes and profound affect resonating within each cautiously crafted phrase. Within the depths of this touching review, we shall discover the book is main harmonies, analyze their enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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 $2 \text{ CO}_2(g) + \text{H}_2\text{O}(g) + 2 \text{ C}_2\text{H}_2(g) \rightarrow 2 \text{ H}_2\text{O}(g) + 2 \text{ C}_2\text{H}_6(g)$   $\Delta h = 94.5 \text{ kJ}$   
 $\text{H}_2\text{O}(g) + \frac{1}{2} \text{ O}_2(g) \rightarrow \text{H}_2\text{O}(l)$   $\Delta h = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(g) + \frac{7}{2} \text{ O}_2(g) \rightarrow 2 \text{ CO}_2(g) + 3 \text{ H}_2\text{O}(g)$   $\Delta h = 283 \text{ kJ}$   
 $\text{C}_2\text{H}_6(g)$

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 $\text{C}_2\text{H}_4(g) + 3 \text{ O}_2(g) \rightarrow 2 \text{ CO}_2(g) + 2 \text{ H}_2\text{O}(l)$   $\Delta h = 1411 \text{ kJ/mole}$   
 $\text{C}_2\text{H}_6(g) + 7/2 \text{ O}_2(g) \rightarrow 2 \text{ CO}_2(g) + 3 \text{ H}_2\text{O}(l)$   $\Delta h = 1560 \text{ kJ/mole}$   
 $\text{H}_2(g) + 1/2 \text{ O}_2(g) \rightarrow \text{H}_2\text{O}(l)$   $\Delta h = 285 \text{ kJ/mole}$

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web thus for reaction 9 2 in which 1 mole of  $h_2o$  and 1 mole of  $oh^-$  combine to form 2 moles of  $h_2o$   $\Delta h = -55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta h_{products} - \Delta h_{reactants}$  is positive if heat is absorbed that is if  $\Delta h_{products}$

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