

18.4 Molecular Structure: Mass

Multiple Choice:

1. The molar masses of quantities in chemical equations are based _____
 (A) percent composition (B) percent yield
 (C) molecular weight (D) stoichiometric coefficients
2. Percent of hydrogen in $\text{C}_2\text{H}_6\text{O}$ is approximately _____ of the molecular weight?

(A) 10% (B) 15% (C) 20% (D) 25%

3. In a gas the other reactions represent oxygen mixed with other elements, 2.5 g of gas was produced by the reaction, and only produced 1.5 g of gas. What is the percent yield of sulfur in this reaction?

(A) 100% (B) 60% (C) 40% (D) 20%

4. In this reaction $2\text{CO} + \text{O}_2 \rightarrow 2\text{CO}_2$, what is the percent yield of oxygen?
 (A) 100% (B) 50% (C) 25% (D) 12.5%

5. What is the percent yield of oxygen in the following equation?



6. Aluminum reacts with H_2 to produce hydrogen gas. If 10.0 g of aluminum reacts with H_2 to produce 1.0 g of H_2 , what is the percent yield?



7. In a full experiment, structure-molecular weight of H_2SO_4 is 98.08 g/mol. If 10.0 g of H_2SO_4 is used and 1.0 g of H_2SO_4 is produced, what is the percent yield of H_2SO_4 in this reaction?



Multiple Choice Stoichiometry Test

Marcel A. Müller



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web nastavni program iz predmeta sociologija iv razred 2 sata sedmično 72 sati godišnje red broj naziv programskih sadržaja
sati 1 tema ili oblast sociologije kao opće teorijske discipline o društvu 1 1 defini cija predme ta sociolo gi je 1 2 odnos pri
rodnih i društvenih pojava prirodnih nauka i društvenih disciplina 1 3