

## 16.5

## HYDROGEN IONS AND ACIDITY

## Section Overview

## Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH values and concentrations of acids and bases and vice versa.
3. Calculate the pH of weak acid solutions.

## Vocabulary

- |                   |                              |                    |
|-------------------|------------------------------|--------------------|
| 1. autoionization | 2. conjugate acid of a base  | 3. pH              |
| 4. autoionization | 5. conjugate base of an acid | 6. polyprotic acid |
|                   | 7. weak acid                 |                    |

## Key Equations

- |                                    |                                                                                  |
|------------------------------------|----------------------------------------------------------------------------------|
| 1. $\text{pH} = -\log[\text{H}^+]$ | 2. $\text{pH} = \text{pK}_a + \log\left(\frac{[\text{A}^-]}{[\text{HA}]}\right)$ |
| 3. $\text{pH} = -\log[\text{H}^+]$ | 4. $\text{pH} = \text{pK}_a + \log\left(\frac{[\text{A}^-]}{[\text{HA}]}\right)$ |

## Part A: Completion

Use the appropriate number to identify the relationship of the conjugate acid-base pair indicated in the second step of the acid-base reaction shown in the following equation.

- \_\_\_\_\_ When phosphoric acid \_\_\_\_\_ reacts with hydroxide ion, the conjugate base \_\_\_\_\_ of the conjugate acid \_\_\_\_\_ is produced in the reaction.
- \_\_\_\_\_ Phosphoric acid reacts with hydroxide ions \_\_\_\_\_ to form the dihydrogen phosphate \_\_\_\_\_ ion.
- \_\_\_\_\_ Hydroxide \_\_\_\_\_ ion reacts with \_\_\_\_\_ to form the conjugate acid \_\_\_\_\_.
- \_\_\_\_\_ \_\_\_\_\_ reacts to form the conjugate base \_\_\_\_\_.
- \_\_\_\_\_ \_\_\_\_\_ reacts to form the conjugate acid \_\_\_\_\_.
- \_\_\_\_\_ \_\_\_\_\_ reacts to form the conjugate base \_\_\_\_\_.

- |     |                |
|-----|----------------|
| 1.  | conjugate acid |
| 2.  | conjugate base |
| 3.  | acid           |
| 4.  | base           |
| 5.  | acid           |
| 6.  | base           |
| 7.  | acid           |
| 8.  | base           |
| 9.  | conjugate acid |
| 10. | conjugate base |
| 11. | acid           |

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**Judd E. Hollander**



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